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PAPER

Surface charge transfer doping of germanium nanowires by MoO₃ deposition

Lin-Bao Luo,^{*a} Tsz-Wai Ng,^b Hao Tang,^b Feng-Xia Liang,^b Yu-Cheng Dong,^b Jian-Sheng Jie,^{*a} Chun-Yan Wu,^a Li Wang,^a Zhi-Feng Zhu,^a Yong-Qiang Yu^a and Qiang Li^a

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We report on the surface transfer p-type doping of germanium nanowires (GeNWs) *via* MoO₃ thin film deposition. The GeNWs studied were prepared by a conventional thermal evaporation approach. Electrical property analysis shows that the conductance, hole mobility and concentration were all prominently enhanced after MoO₃ thin film coating. Such a remarkable surface doping effect can be attributed to the surface charge transfer at the GeNWs/MoO₃ interface, which is verified by *in situ* XPS analysis of GeNWs as a function of increasing MoO₃ coverage. Further hole mobility and concentration evolution study of MoO₃/GeNWs reveals that the GeNWs embedded in the MoO₃ layer can retain their electrical property after storage in air for 3 months. The generality of the above results suggests that the charge transfer doping *via* surface deposition has great potential in GeNW-based nanoelectronic devices and may also be applicable to the modulation of other semiconductor nanostructures.

Introduction

In the last decade, one-dimensional (1D) semiconductors, such as group IV, III–V, and II–VI compounds nanowires (NWs), have been actively investigated in an effort to utilize them for various advanced applications, *e.g.*, field-effect transistors (FETs),^{1,2} logic circuits,^{3,4} light-emitting diodes (LEDs),^{5,6} solar cells,^{7,8} photodetectors,^{9,10} and biological or chemical sensors.^{11,12} In particular, elementary semiconductor NWs, such as silicon NWs (SiNWs) and germanium NWs (GeNWs), have exhibited unique electrical properties with respect to their bulk counterparts and good compatibility with the existing semiconductor technology. SiNWs and GeNWs are the most promising building blocks for future functional devices at the nanoscale level.¹³

Compared with Si, Ge has higher electron and hole mobilities, a lower intrinsic resistivity and larger Bohr exciton radius.^{13,14} This unparalleled superiority, combined with the convenient synthesis renders GeNWs appealing candidates for low drive voltage, high drive current and high-speed electronic devices.^{15,16} Nonetheless, undeniably, due to the huge surface to volume ratio, the electrical properties of GeNWs are highly dependent on the surrounding atmosphere and difficult to control. On the other hand, GeNWs can easily get oxidized in ambient conditions, forming a water soluble GeO₂ sheath surrounding the NW.¹⁷ This feature makes GeNWs highly unstable and thereby constitutes the major obstacle to their wide application

in optoelectronic devices.¹⁸ Exploration of effective ways to rationally control the electrical properties of GeNWs with favorable stability is thus highly desirable.

The development of the surface charge transfer technique holds great promise. As established previously, surface doping is completely different from the traditional volume doping method in that it can allow direct charge transfer from one material to another by virtue of their remarkable differences in work functions.¹⁹ Moreover, compared with the traditional doping method by intentionally introducing impurity atoms into the crystal lattice, this surface-induced doping approach is relatively convenient and can allow rational tailoring of the electrical properties of semiconductor without damaging the interior structures. In this study, we present an attempt to dope GeNWs *via* a surface charge transfer approach. The doping effect of the GeNWs through MoO₃ coating were evaluated by characterizing single GeNW-based FETs and the surface charge transfer at MoO₃/GeNW interface was probed by *in situ* X-ray photoelectron spectroscopy (XPS). The observed considerable injection of holes from MoO₃ into GeNW, along with the long-term stability suggests that the surface charge transfer doping is applicable to the modulation of the electrical properties of other semiconductor nanostructures.

Experiment

Synthesis and characterization of germanium nanowires

The fabrication of GeNWs was carried out on a three zone horizontal tube furnace.¹⁹ Briefly, 2 g of germanium powder (99.99%, Sigma-Aldrich Co.) was placed in an alumina boat at the center of the tube furnace. Then, a piece of thermally oxide p⁺-Si wafer (300 nm SiO₂) loaded with a very thin layer of gold

^aSchool of Electronic Science and Applied Physics, Hefei University of Technology, Hefei, Anhui, 230009, P. R. China. E-mail: luolb@hfut.edu.cn (L. B. Luo); jsjie@hfut.edu.cn (J. S. Jie)

^bDepartment of Physics and Materials Science, City University of Hong Kong, Kowloon, Hong Kong SAR, P. R. China

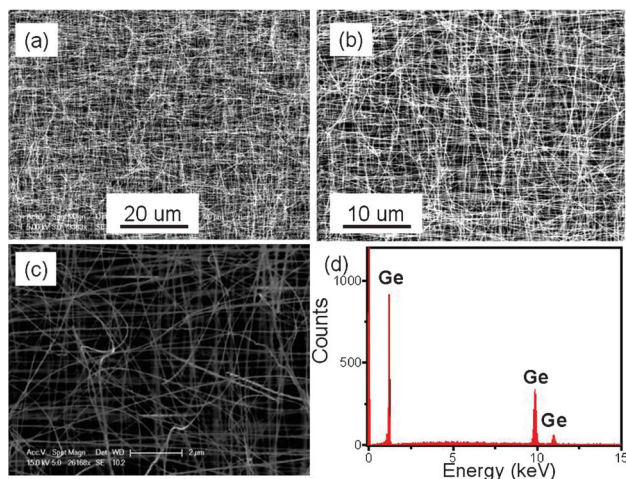


Fig. 1 (a–c) SEM images of the GeNWs at different magnification. (d) The corresponding EDX spectrum.

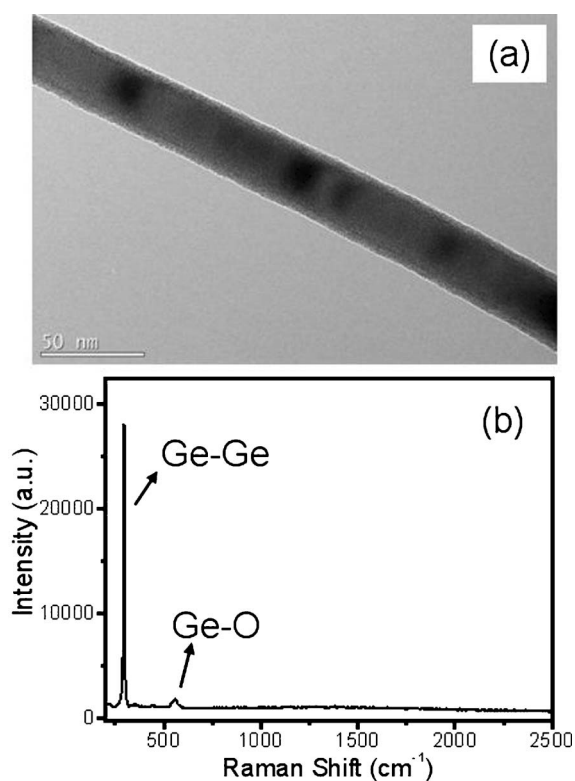


Fig. 2 (a) A TEM image of a typical GeNW. (b) The Raman spectrum of GeNWs showing the presence of both Ge and Ge oxide.

nanoparticles was placed in the downstream direction. After the tube was pumped down to a pressure of 1×10^{-2} mbar, high purity mixed gas of H_2/Ar (5/95%) was introduced into the tube at a rate of 20 sccm. The germanium powder was increased to 900°C at a rate of $15^\circ\text{C min}^{-1}$ and the Si substrate to 450°C at a rate of $7.5^\circ\text{C min}^{-1}$, and kept for about 1 h. Throughout the growth process, the system was maintained at 200 mbar. Finally, the product was collected from the SiO_2/Si substrate after the system was allowed to cool down to ambient temperature. The

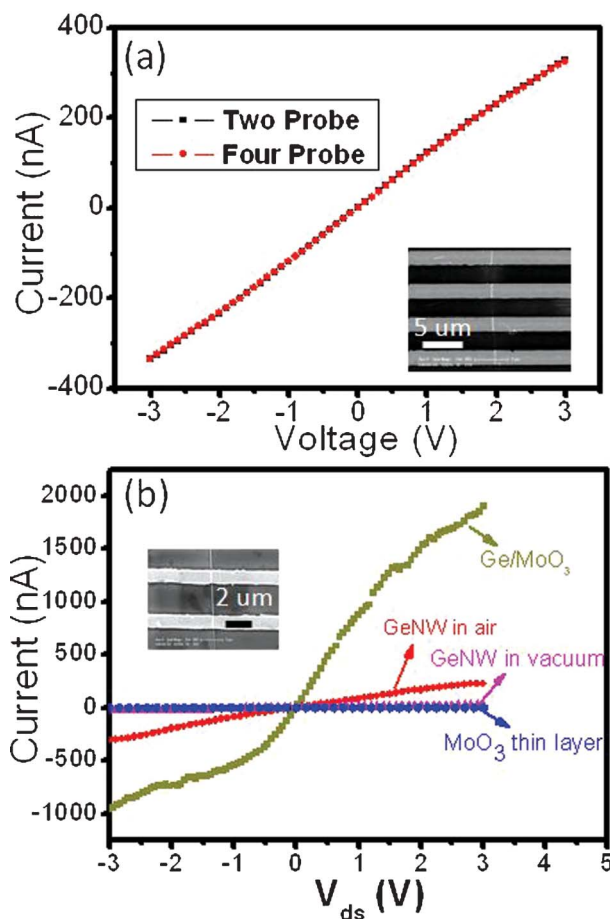


Fig. 3 (a) I - V curves measured in both two-probe and four-probe configurations; inset shows four parallel Ti/Au electrodes bridged by an individual GeNW. (b) Conductance of GeNW in air, vacuum, GeNW coated with MoO_3 (GeNW/ MoO_3), and MoO_3 thin film, inset is the SEM image of a typical GeNW-FET.

morphologies and crystal structures were investigated by field-emission scanning electron microscopy (FESEM, Philips XL 30 FEG) and transmission electron microscopy (TEM, Philips CM20, operating at 200 kV). The elemental compositions were analyzed by energy dispersive spectroscopy (EDS) attached to the FESEM system. The Raman scattering measurements were performed on a Renishaw 2000 spectrometer with incident 633 nm laser.

Device fabrication and measurement

The fabrication of single-GeNW based FETs followed a standard photolithography process. Briefly, the as-prepared GeNWs were first scraped from the Si substrate and immersed into an absolute alcohol solution *via* ultrasonication. The resulting NWs suspension was then spread onto a 500 nm $\text{SiO}_2/p^+\text{-Si}$ substrate. Photolithography (Mask aligner: Karl Suss MJB-3) and a lift-off process were employed to define the Ti (2 nm)/Au (80 nm) source and drain electrodes with 2 μm finger spacing on the individual GeNWs. Afterwards, 100 nm Al was deposited onto the back of the $p^+\text{-Si}$ substrate, which serves as the back gate electrode. To study the effects of surface transfer doping, MoO_3 thin film was directly deposited onto the GeNW-based FETs

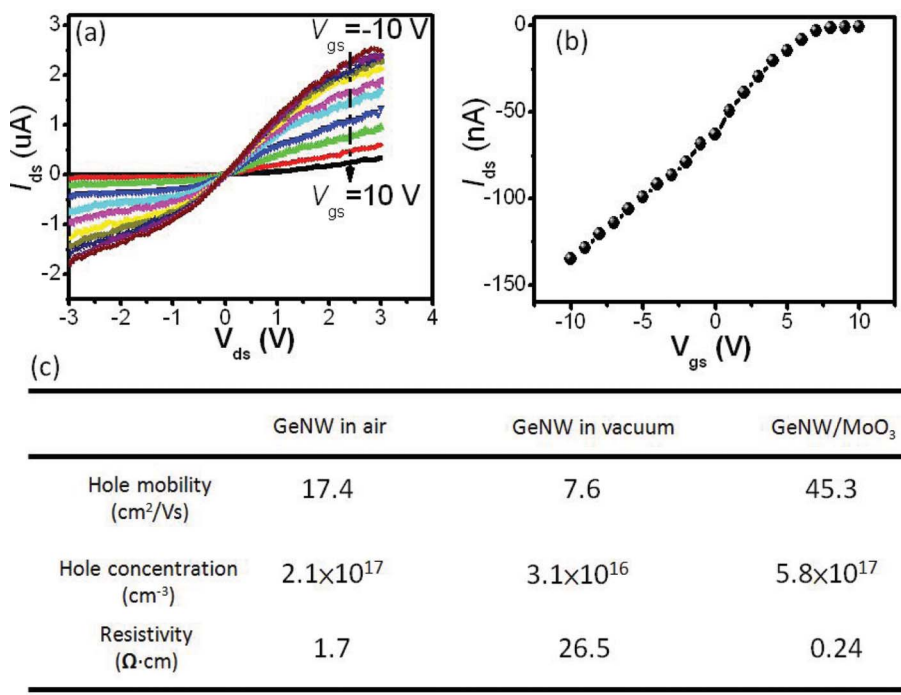


Fig. 4 (a) I_{ds} - V_{ds} curve of GeNW after deposition of MoO₃, under different gate voltages. (b) Corresponding transport characteristics at $V_{ds} = -0.1$ V. (c) Table of electrical properties of GeNW in air, vacuum and with a layer of MoO₃ thin film.

through a thermal evaporator. The evaluation of the GeNW FETs was carried out on a semiconductor characterization system (Keithley 4200-SCS).

XPS analysis

XPS analysis was performed *in situ* using a VG ESCALAB 220i-XL surface analysis system equipped with a monochromatic Al K α X-ray (1486.6 eV) source. The analysis system is composed of interconnected analysis, coating and preparation chambers with vacuum pressures of 3×10^{-10} , 5×10^{-10} , and 3×10^{-7} Torr, respectively. The MoO₃ layer was deposited stepwise by thermal evaporation inside the deposition chamber at pressure of around 1×10^{-9} Torr. The freshly prepared samples were transferred to the analysis chamber for *in situ* XPS study without vacuum break.

Results and discussion

The general overview FESEM image in Fig. 1(a) shows that the product obtained by evaporating pure Ge powder is composed of network-like NWs. The GeNWs are up to tens of micrometres in length, and exhibit high abundance and purity without any appreciable by-products. Further magnified FESEM and TEM images in Fig. 1(b) and (c) reveal that the NWs are comparatively uniform along the whole length with diameter in the range 40–90 nm. In addition to Ge which is found in the elemental composition analysis shown in Fig. 1(d), a layer of amorphous germanium oxide thin film (GeO₂ or GeO) was observed on the GeNWs surface (*cf.* the XPS result in Fig. 6). This species, according to the early work, results from the rapid oxygen oxidation when exposed to air, which is usually unavoidable during NWs preparation.²⁰ In fact, the presence

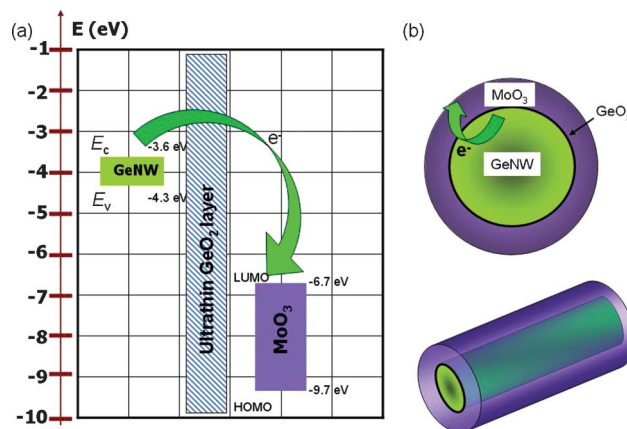


Fig. 5 (a) A schematic illustration of the electron injection from GeNW to MoO₃. (b) An electronic structures diagram of both GeNW and MoO₃.

of a thin Ge oxide layer is also confirmed by the Raman spectrum shown in Fig. 2(b).²¹

Fig. 3(a) shows the typical I - V curves of a single GeNW acquired from both two-probe and four-probe approaches. The linear curves throughout the voltage measurement range reveal the good ohmic contact between the electrodes and GeNW. The resistance of the NW is calculated to be $8.9 \times 10^6 \Omega$ and the contact resistance, namely the difference between the two-probe and four-probe resistance is $4.1 \times 10^4 \Omega$. In light of the relatively low contact resistance, probably due to the matched work function,²² a two-probe method is used in the following experiments. The inset in Fig. 3(b) shows a typical FESEM image of the GeNW FET after 20 nm MoO₃ deposition, in which two

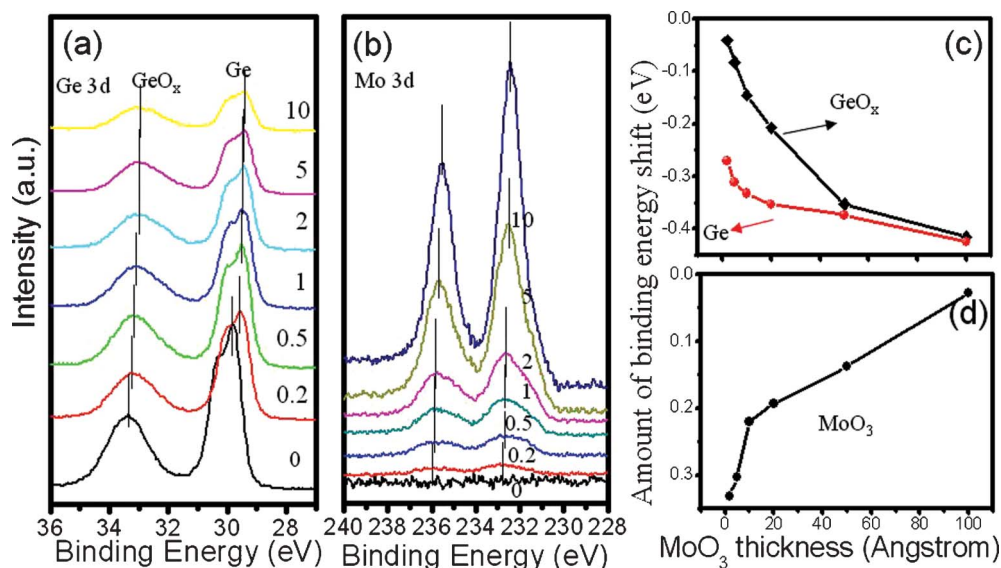


Fig. 6 XPS spectra of Ge 3d (a) and Mo 3d (b) peaks for GeNWs coated with MoO₃ of increasing thickness of 0, 0.2, 0.5, 1, 2, 5 and 10 nm. The binding energy shift of Ge 3d for GeNWs (c) and Mo 3d for MoO₃ (d) with increasing MoO₃ coverage. The shift of the binding energy confirms the surface charge transfer at the GeNWs and MoO₃ interface.

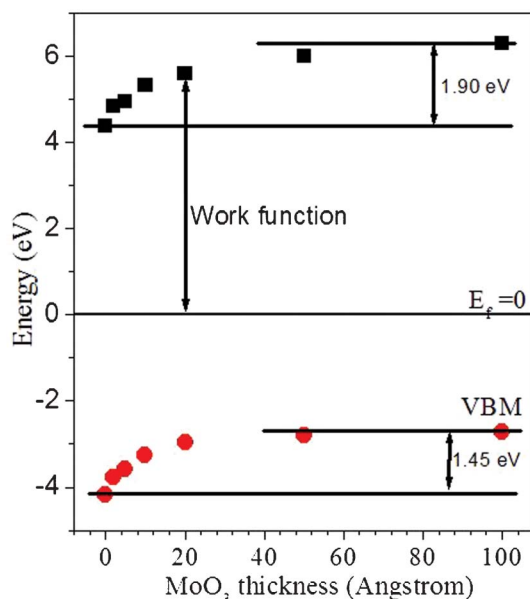


Fig. 7 The energy level alignment of MoO₃ as a function of increasing MoO₃ thickness.

parallel gold electrodes are crossed by an individual GeNW. The effective gate length is 2.5 μm and the diameter of the NW is 80 nm. Fig. 3(b) plots I_{ds} vs. V_{ds} curves of a pure MoO₃ thin layer and GeNW in air, at vacuum and after MoO₃ coating. Among the four structures, MoO₃ shows the lowest conductance, while the GeNW has higher conductance in air than when in a vacuum, mainly due to the ambient effect. Interestingly, after coating with a 20 nm MoO₃ thin layer, the as-obtained GeNW/MoO₃ structure shows the highest conductance. These results suggest that the substantial increase of the conductance for the MoO₃-coated GeNW (GeNW/MoO₃) cannot be solely attributed to the conductance enhancement of the MoO₃ coating layer.

As we will discuss later, the surface charge transfer between the MoO₃ and GeNW interface is likely responsible for this phenomenon.

Further electrical transport characterization indicates that the undoped GeNWs without MoO₃ coating shows weak p-type semiconductivity (data not shown). This behavior is attributed to the presence of a huge quantity of surface dangling bonds that can lead to the formation of an acceptor level slightly above the valance band maximum.²³ In contrast, the p-type conductivity of the GeNW is remarkably enhanced after coating with MoO₃ thin layer. Fig. 4(a) and (b) depict the transport characteristics of the GeNW FET after MoO₃ coating. The device exhibits a typical behavior of a p-channel semiconductor FET, with a hole mobility and concentration of 45.3 $\text{cm}^2 \text{V}^{-1} \text{s}^{-1}$ and $5.8 \times 10^{17} \text{cm}^{-3}$, respectively. A comparison of the hole mobility, hole concentration and resistivity of the GeNW in air, a vacuum and after MoO₃ deposition is summarized in Fig. 4(c). It is worth noting that owing to oxygen and water molecules absorption on the surface dangling bonds, an acceptor level above the valance band was formed, contributing to the p-type conductivity of the GeNW in air. Desorption of these surface molecules in a vacuum leads to the decrease of the hole concentration. As a result, the resistivity of the GeNW increases by nearly one order of magnitude in a vacuum. On the other hand, the MoO₃ coating on the GeNW results in the increase of both the hole mobility and the hole concentration. Consequently, the GeNW resistivity decreases by one order of magnitude with respect to that in air. The enhanced carrier mobility of GeNW after MoO₃ coating is probably due to the reduced carrier scattering as a consequence of surface oxygen and water molecules desorption, as well as the effective surface passivation.^{24,25} As for the increase in hole concentration, it is assumed to be associated with the mutual surface charge transfer between the GeNW and MoO₃ thin layer due to their differences in work function, which will be verified by the *in situ* XPS analysis result later on.

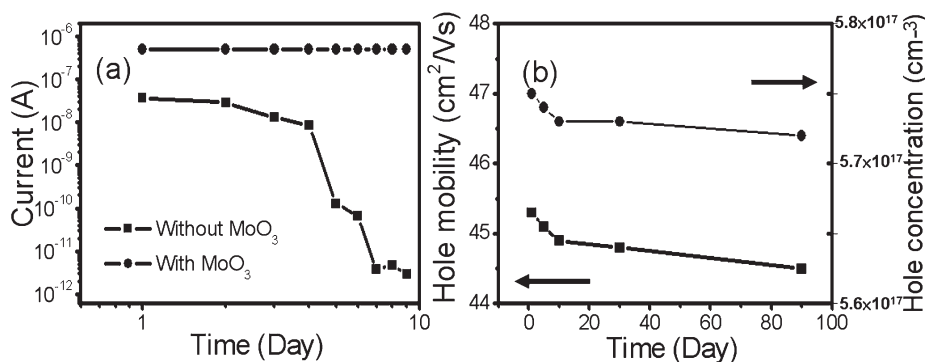


Fig. 8 (a) Current evolution of a GeNW with and without MoO₃ thin film deposition. (b) Hole mobility and concentration evolution of the MoO₃-coated GeNW as a function of time.

Previously, it has been reported that MoO₃ has a very low lowest unoccupied molecular orbital (LUMO) value of -6.7 eV and highest occupied molecular orbital (HOMO) value of -9.7 eV,²⁶ signifying that the work function of MoO₃ is comparatively larger than the majority of other materials including organic and inorganic semiconductors, thereby favoring holes injection from MoO₃ to the other semiconductors, which render it an effective hole injector in organic LEDs and organic FETs.^{27,28} In our case, the huge difference in work functions between MoO₃ and GeNW also accounts for the enhanced p-type conductivity in GeNW after MoO₃ coating. As we can see in Fig. 5(a), the conduction band minimum (E_c) and the valence band maximum (E_v) of the GeNW are located at -3.6 and -4.3 eV, respectively. The Fermi level is above both the HOMO and LUMO of MoO₃. As a result, electrons will transfer spontaneously from GeNW to MoO₃ (Fig. 5(b)), giving rise to the increase in hole concentration within the GeNW. It is noteworthy that although an ultrathin GeO₂ sheath is often observed on the outside of the GeNWs, it is too thin to block the electron transfer *via* direct tunneling. Moreover, similar electron transfer from GeO₂ to MoO₃ was also investigated in the XPS analysis below.

As a matter of fact, the observed surface charge transfer doping effect was experimentally verified by XPS spectrum evolution of Ge 3d and Mo 3d with increasing nominal MoO₃ coverage, as shown in Fig. 6(a). It can be clearly visualized that prior to MoO₃ coating, the spectrum of Ge 3d is mainly composed of two Ge 3d peaks centered at binding energy of 29.84 and 33.32 eV, corresponding to Ge and GeO₂, respectively. When the MoO₃ thin layer is deposited stepwise, a gradual shift of Ge 3d towards low binding energy is observed for both peaks. Fig. 6(c) plots the binding energy shift as a function of MoO₃ layer thickness, in which a total shift of about 0.43 eV for Ge and 0.42 eV for GeO₂ is observed as MoO₃ thickness increase from 0 to 10 nm. This binding energy shift corresponds to the upward energy level bending from bulk to interface, indicating that GeNW loses electrons near the heterojunction interface. The surface charge transfer effect at the GeNW/MoO₃ interface is corroborated by the corresponding Mo 3d spectrum evolution as well. As shown in Fig. 6(b), with the increase in MoO₃ thickness, the binding energy of Mo 3d shifts downwards by 0.33 eV; this shift suggests that the electrons were injected from the GeNW to MoO₃, in good agreement with the enhanced p-type electrical conduction behavior.

The charge transfer at the interface is also verified by the formation of an interface dipole. Fig. 7 plots the evolution of work function and VBM (*i.e.*, the band bending) of MoO₃ as a function of MoO₃ layer thickness, from which one can see that as the dopant layer thickness increases from 0 to 100 Å, both the work function and VBM increase. Notably, a careful comparison of both values concludes that the change in work function is always larger than that of the band bending. This result suggests the existence of an interfacial dipole at the MoO₃/GeNW interface, thus corroborating the electron transfer from GeNW to MoO₃ at the contacting junction, considering the fact that the overall change in work function includes the contribution from both band bending and interfacial dipole.^{29,30}

It is worth mentioning that the MoO₃ thin layer can not only serve as an effective hole injector, but also help prevent the GeNWs from getting oxidized when exposed in ambient conditions, which is often observed on GeNWs and responsible for the degeneration of their electrical properties.³¹ As depicted in Fig. 8(a), the GeNW without MoO₃ thin layer coverage can get easily oxidized and its electrical conduction attenuates at a considerable pace in less than 1 week. However, the deposition of a MoO₃ thin layer can act as a protective layer and greatly restrict such an oxidation process. Fig. 8(b) shows the hole mobility and the hole concentration of a 20 nm MoO₃-coated GeNW FET as a function of exposure duration, from which one can see that the hole mobility decreases slightly from 45.3 to 44.5 cm² V⁻¹ s⁻¹, and hole concentration decreases slightly from 5.75×10^{17} to 5.72×10^{17} cm⁻³ after storage in air for 3 months. The change in the device parameters is less than 2%, which unambiguously demonstrates that the GeNWs embedded in the MoO₃ thin layer are extremely stable, greatly facilitating their applications in future nanoelectronic devices.

In summary, we have presented a surface transfer approach to the effective p-type doping of GeNWs *via* MoO₃ thin film coating. The GeNWs studied were prepared by a conventional thermal evaporation approach. The electrical property study of the single GeNW-based field effect transistor and *in situ* XPS analysis of GeNWs as a function of increasing MoO₃ coverage support the supposition that a considerable quantity of holes was transferred from MoO₃ to GeNWs due to their difference in work function. Finally, the long-term electrical property study of the GeNW reveals that the hole mobility and concentration remain almost the same. The generality of this work demonstrates that surface

transfer doping is an effective method to tailor the electrical property of GeNWs.

Acknowledgements

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